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$2 \text{ (g)} + 2\text{H}_2\text{O} \text{ () } \rightarrow \text{CH}_4 \text{ (g)} + 2\text{O}_2 \text{ (g)}$ During the reaction the temperature of the reaction mixture decreases. 6.1 Define the term enthalpy change. (2) 6.2 Does the enthalpy change (ΔH) for this reaction have a positive or negative value? Explain the answer by referring to the energy involved . (2)

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