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nature of both real and ideal  
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Answers of gases that can be used to explain their characteristics: volume, number of particles, temperature, and pressure.

Chapter 13 - Gases - An  
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Answers

Chemistry Answers 1. Gases consist of tiny particles (atoms or molecules) 2. These particles are so small, compared with the distance between them, that the volume (size) of the individual particles can be assumed to be negligible (zero) 3. The particles

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A gas is a state of matter with no defined shape or volume. Gases have their own unique behavior depending on a variety of variables, such as temperature,

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Answers  
pressure, and volume. While each gas is different, all gases act in a similar matter. This study guide highlights the concepts and laws dealing with the chemistry of gases.

Chemistry Study Guide for Gases

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STUDY GUIDE AP Chemistry  
Chapter Ten- Gases Lecture Notes  
101 Characteristics of Gases □All  
substances have three phases:  
solid, liquid and gas □Substances  
that are liquids or solids under  
ordinary conditions may also exist

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Answers These are often referred to as vapors Many of the properties of gases differ

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number of particles of a gas are constant, which of the following is also constant? ... Chemistry Chapter 13-15 study guide. 45 terms. hmcoates. quiz 5. 26 terms. savannahlyynn.

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substances have three phases:  
solid, liquid and gas □Substances  
that are liquids or solids under

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## Problems Answers

STUDY GUIDE AP Chemistry

Chapter Ten- Gases Lecture Notes

10.1 Characteristics of Gases □All

substances have three phases:

solid, liquid and gas. □Substances

that are liquids or solids under

ordinary conditions may also exist

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Answers. □ These are often referred to as vapors. □ Many of the properties of gases differ from those of solids ...

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gases. 2. What 3 factors affect  
gas



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