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washers $0.110 \text{ g} \times 1 \text{ washer} = 11.0 \text{ g}$
(assuming 100 washers is exact)
 $100. \text{ g} \div 1 \text{ washer} \times 0.110 \text{ g} = 909$
washers $2.500. \text{ g} \div 1 \text{ cork} \times 1.63 \text{ g} =$
 $306.7 = 307$ corks $500. \text{ g} \div 1$
stopper $4.31 \text{ g} = 116$ stoppers 1
kg (1000 g) of corks contains
(1000 g

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Interactions; 2 Take 5 _____ is the
condition that results from changes
in the independent variable. _____

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results in an increase in the
amount of living material and the
formation of new structures. 3

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6. (c) 7. (d) 8. (b) 9. (a) 10. (c)

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Questions. Question 1. Why standard entropy of an elementary substance is not zero whereas the standard enthalpy of formation is taken as zero? Answer: A substance has a perfectly ordered

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Arrangement only at absolute zero.
Hence, entropy is zero only at ...

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deals with exchange of energy in the form of heat and temperature. William Thomson coined this term in 1794 by combining two Greek words Thermes which is heat and Dynamikos meaning powerful.

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211 the reaction we monitor the
mass of Ca^{2+} remaining in
solution and the mass of CaCO_3
that precipitates, the result looks
something like Figure 6.1. At the

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start of the reaction the mass of Ca^{2+} decreases and the mass of CaCO_3 increases. Eventually the reaction reaches a point after which there

Chapter 6

Chapter Chapter 6: Speaking of

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