

Chemistry Chapter 3 Test Review

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Chemistry Chapter 3 Test Review. a system of physical units (SI Units) based on the meter, kilogram, second, ampere, kelvin, candela, and mole, together with a set of prefixes to indicate multiplication or division by a power of ten.

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Chemistry - Chapter 3 Test Review Answer all the questions in complete sentences (just restate the original question). Show the math work for # 27-30. 1. The law of conservation of mass follows from the concept that _____. 2. If 4 g of element A combine with 10 g of element B, then 12 g of element A combine with _____ g of element B. 3.

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Chapter 3 Chemistry Review. dalton's five principles of the atomic.... law of conservation of mass. law of multiple proportions. law of definite proportions. 1. all matter is composed of extremely small particles called.... mass is neither created nor destroyed in a chemical reaction b....

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Chapter 3 Chemistry Test Review. anything that has mass and takes up space. matter composed of all one type of atom (can be a single element, like He or a diatomic element, like O₂ or H₂). Elements cannot be broken down into simpler substances. chemical combinations of two or more elements.

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Review Skills. The presentation of information in this chapter assumes that you can already perform . the tasks listed below. You can test your readiness to proceed by answering the Review . Questions at the end of the chapter. This might also be a good time to read the Chapter . Objectives, which precede the Review Questions.

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Holt McDougal Modern Chemistry 3 Chapter Test Chapter Test B, continued 16 Modern chemistry chapter 3 test b answers. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called _____. 17. The energy required to remove one electron from an atom is called its _____. 18.

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