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Volume \u0026amp; Mass

Percent, Mole

Fraction \u0026amp;

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Solutions

The molal freezing

point constant, K_f , for

water is $1.86 \text{ } ^\circ\text{C kg} /$

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mol. A solution of 0.5 g of an unknown nonvolatile, nonelectrolyte solute is added to 100 mL of water and then placed across a ...

Colligative Properties of Solutions

In nonelectrolyte solutions, heat effects are usually endothermic and

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much smaller, often about 100 calories, when roughly equal parts are mixed to form 100 grams of mixture. Formation of a solution ...

Weak electrolytes

For example, when dinitrogen pentoxide is dissolved in water, a new substance, nitric acid, is formed;

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and it is necessary to interpret the behaviour of such a solution in terms of its chemical ...

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