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[ELEC 353 - Assignment #8 2nd part .](#) Input line quarter-wave transformer line #3 . 1. An antenna operating at 1900 MHz has input impedance . $Z_L = 90 - j40$ ohms. The matching circuit shown above has an input line of length 5 cm. The quarter wave transformer has length L_t and line #3 has length L . The characteristic resistance of the input line and of line #3 is R_c and the characteristic ...

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[ELEC 353 - Solution to Assignment #9 2.](#) A plane wave travels in the z direction in a material at 850 MHz. The relative permittivity of the material is $\epsilon_r = 9$ and the loss tangent is 0.15. The electric field is oriented parallel to the x axis. The amplitude of the electric field at $z=0$ is 5 volts/meter. $\omega = 2\pi \cdot f = 5.3407 \times 10^9$ rad/sec

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[ELEC 353 - Solution to Assignment #5](#) This assignment is included on the class test. 1. A step-function voltage source with $V_s = 10$ volts and internal resistance $R_s = 50$ ohms drives a transmission line having characteristic resistance $R_c = 50$ ohms and speed-of-travel $u = 14$ cm/ns. The line is 3 cm long. The load terminating the line has pin inductance $L = 2$ nH in series with a chip input resistance ...

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[ELEC 353 - Assignment #1 Note:](#) In ELEC353, assignments are not handed in. Do the assignments week-by-week and then evaluate your work in comparison to the solution. 1. What is an " ideal " short circuit? 2. In the circuit shown above, s V and s R

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Page 1 of 4 [ELEC 216 - Electricity and Magnetism Name ____ Assignment 8](#) Due on Thursday, 30 July 2020, 6:00 pm in Course Spaces Std # ____ SOLUTIONS Problem 1 A coaxial line with inner and outer conductor diameters of 0.5 cm and 1 cm, respectively, is filled with an insulating material with $\epsilon_r = 4.5$ and $\sigma = 10^{-3}$ S/m. The conductors are made of copper.

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[ELEC 2220 - Homework Assignment #1 Prerequisite Review](#) Due Wednesday, May 20, 2015 1. Write and execute a short computer program that reads non-zero integer numbers, entered from a keyboard, until a zero is entered, and then identify and print the smallest number, the largest number, and the average value (don ' t count the final 0 in the average).

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[ELEC 3400 Homework Assignment #1](#) May 27, 2010 (Class discussion on June 2) 1. Problem 7.3 in Haykin & Mohler. 2. (a) A signal $m(t) = 2\cos(8000t)$ is sampled at the rate of 6000 samples/sec.

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[ELEC 3400 Homework Assignment #7](#) April 11, 2005 (Class discussion on April 15, 2005.) 1. Using a unit amplitude, 50% roll-off raised cosine pulse to indicate a 1 and no pulse to indicate a 0, sketch the pulse sequence corresponding to the binary message sequence 11001. - 1068345

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With platinum and rhodium, palladium is one of the most important members of the platinum metal group. The last Gmelin treatment of it was in 1942, and knowledge of its properties and chemistry has made enormous strides since then. This volume is primarily concerned with binary compounds and with the coordination complexes derived from them. Although it is a member of the nickel-palladium-platinum triad, it more closely resembles platinum in its binary and coordination chemistry, though being a second-row transition element it displays less tendency than does platinum to assume higher oxidation states. In heterogeneous and homogeneous catalysis, referred to at appropriate points, palladium and its complexes are of great importance in bulk and fine chemicals production, effecting a wide variety of organic transformations. The arrangement of material in this volume follows the traditional Gmelin arrangement. Within each category of compounds or complexes the material is arranged, as usual, in order of ascending metal oxidation states (e. g., palladium(II) precedes palladium(IV)). The chemistry of the palladium-hydrogen system is so large that it merits a separate volume, so this book starts with the binary oxides and oxopalladates followed by hydroxides, hydroxo complexes and aquo complexes. Then nitrides and nitrates are treated. They are followed by the large chapters on halides and their complexes (172 pages). The largest single chapter in this volume (11 0 pages) deals with chlorides, chloropalladates and other chloro complexes.

The present volume describes cyclic sulfur-nitrogen compounds whose ring system contains one or two other types of atoms in addition to sulfur and nitrogen. At least one sulfur atom can be regarded as having oxidation number IV in one resonance structure. That means the sulfur in fact has an oxidation number higher than 11 and lower than VI. The volume continues "Sulfur-Nitrogen Compounds" Part 2, in which the binary sulfur(IV)-nitrogen ring systems are described. Compounds with the same "hetero atom" in the sulfur-nitrogen ring are arranged in groups. Within a group the compounds are arranged according to ring size, and for a given number of ring atoms, in order of decreasing S: N ratio. Neutral compounds are described before ions, and saturated compounds before unsaturated, aromatic ones. A very heterogeneous material is brought together in this volume. SN Se2 rings, which 2 strongly resemble the binary S3N2 ring, are covered. S-N-Si rings with five, six, and eight atoms, and also N-S-N-bridged eight-membered rings, are described. The S-N-P rings form a puzzling and complex part of this volume; they readily interconvert by dimerization, cleavage, and by addition and loss of NSN and NPN groups. Especially interesting is the aromatic character of 8-N-P rings.

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