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Question Answer Marks

Guidance 1 (a) (The enthalpy

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change that accompanies) the formation of one mole of a(n ionic) compound from its gaseous ions (under standard conditions) 2 IGNORE 'energy needed' OR 'energy required' ALLOW as alternative for compound: lattice, crystal, substance, solid

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Guidance Throughout Q1

IGNORE variations in caps

and small letters 1 (a) (i)

Fe 1

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Afternoon A2 GCE CHEMISTRY A
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The Insert will be found in
the centre of this document.

† Write your name, centre
number and candidate number
in the boxes above. Please
write clearly and in capital
letters. † Use black ink.

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Question Answer Marks
Guidance 1 (a) IGNORE
Formation of one mole of a(n
ionic) compound from its
gaseous ions IGNORE standard
conditions 2 'Energy needed'
OR 'energy required' For
'compound', ALLOW: lattice,
crystal, substance, solid
Special case: 1 mark for
gaseous ions ONLY

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in conjunction with the
published question papers
and the report on the
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Answer Marks Guidance 1 (d)
For FIRST TWO marking
points, assume that the
following refer to 'ions',
Mg. 2+, etc.

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Question Expected Answers
Marks Additional Guidance 2
a $\text{BrO}_3^- + 5\text{Br}^- + 6\text{H}^+ \rightarrow 3\text{Br}_2 + 3\text{H}_2\text{O}$ 1 ALLOW multiples b
graph: Straight/diagonal
line through origin OR 0,0
AND 1st order with respect
to BrO_3^- - 1 ANNOTATIONS MUST
BE USED Both explanation and
1st order required for mark

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