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Chapter 16 The Periodic Table - O Level Chemistry tuition ...

The error in print is on Page 271 under the Chapter of " Electrolysis " in Chemistry Matters for Pure Chemistry syllabus. In the Key Ideas which acts like sub-section summary, it states that " OH⁻ ions will be discharged as oxygen gas. However, if a solution is concentrated in halide ions, the halogen gas will be produced. "

Error in Marshall Cavendish GCE O-Level Chemistry Matters ...

Chemistry Matters for GCE 'O' Level (2 nd Edition): Textbook Answers Chapter 9 © 2013 Marshall Cavendish International (Singapore) Private Limited 9.2 (c) Number of moles of C I₂ gas = mass of C I₂ / molar mass of C I₂ = 1.42 / 71 = 0.02 mol (d) Number of moles of sodium chloride = number of ions of sodium chloride / 2 = 6 / 2 = 3 mol

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Test Yourself 4.1 (page 74) 1. (Any four differences) Mass Weight • An amount of matter • A gravitational force • A scalar quantity (i.e. has only magnitude) • A vector quantity (i.e. has both magnitude and direction) • SI unit: kilogram (kg) • SI

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