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Terms in this set (32) Which one of the following reactions is NOT balanced?
 $2 \text{CO} + \text{O}_2 \rightarrow 2 \text{CO}_2$ $2 \text{SO}_2 + \text{O}_2 \rightarrow 2 \text{SO}_3$ $\text{SF}_4 + 3 \text{H}_2\text{O} \rightarrow \text{H}_2\text{SO}_3 + 4 \text{HF}$ $2 \text{KNO}_3 + 10 \text{K} \rightarrow 5 \text{K}_2\text{O} + \text{N}_2$.

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ANSWER: Correct Chapter 9 Reading Question 2 Part A Iron reacts with oxygen in the air to produce rust (Fe_2O_3) according to the reaction shown below.
 $4 \text{Fe} + 3 \text{O}_2 \rightarrow 2 \text{Fe}_2\text{O}_3$
How many moles of rust can be formed if 1.8 mols of iron are present? Hint 1. Begin with your given number of moles.

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Mastering Chemistry Chapter 2: Rank from most to fewest valence electrons of the following elements: Kr, Ga, Br, As, K. - Answer: Kr (8), Br (7), As (5), Ga (3), K (1). Valence electrons are equal to the atom's main group number. Link: <https://www.coursehero.com/file/20333255/Mastering-Chemistry-Chapter-2/>.

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