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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope.

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3 Atoms: The Building Blocks of Matter  
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CHAPTER 9 REVIEW Stoichiometry MIXED REVIEW  
SHORT ANSWER Answer the following questions in  
the space provided. 1. Given the following equation:  $C_3H_4(g) + xO_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$  4 a. What  
is the value of the coefficient x in this equation? 40.07  
g/mol b. What is the molar mass of  $C_3H_4$ ? 2 mol O<sub>2</sub>:1  
mol H<sub>2</sub>O c. What is the mole ratio ...

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Modern Chemistry Chapter 2 Answers  
Vocabulary from Chapter 6 of the Modern Chemistry

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textbook. (30 total words) Terms in this set (30)  
chemical bond. the attractive force that holds atoms or  
ions together. ionic bonding. a force that attracts  
electrons from one atom to another, which transforms a  
neutral atom into an ion.

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