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chemistry is
ripe with
conjectures,
possibilities,
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arguments, and
intelligent
guesses. The
chemical effects

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of cavitation
have been
explained by
means of many
theories,
consisting of
empirical
constants,
adjustable
parameters, and
the like. The
chemists working
with cavitation
chemistry agree

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that the phenomenon is very complex and system specific. Mathematicians and physicists have offered partial solutions to the observed phenomena on the basis of cavitation parameters,

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whereas chemists
have attempted
explanations
based on the
modes of
reaction and the
detection of
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species.

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one has been
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formulate a

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unified theme,
however crude,
for its effects
on the basis of
the known
parameters, such
as cavitation
and transient
chemistry
involving
extremely high
temperatures of
nanosecond
durations. When

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one surveys the
literature on ca-
vitation-
assisted

reactions, it is
clear that the
approach so far
has been

"Edisonian" in
nature. While a
large number of
reactions have
showed either
enhanced yields

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or reduced reaction times, many reactions have remained unaffected in the presence of cavitation. The success or failure of cavitation reactions ultimately depends on the collapse of the

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chemistry is
based on the

principles of
the formation of
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cavities, their
growth and
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